

Environmental Chemistry

Introduction

Dato' Prof. Zaini Ujang

Ph.D., PE (M), C.Eng.(UK), C.W.E.M. (UK), C.Sci. (UK)

Institute of Environmental & Water Resource Management Universiti Teknologi Malaysia

zaini@utm.my

http://www.cheme.utm.my/staff/zaini



Environmental Chemistry

Lecture outline

Pollution perspective

Major pollutants

Effects of pollutants on ecosystems

Fate and behavior of chemicals in environment

Environmental toxicity testing

Environmental monitoring

Instrumentations

ocw.utm.my

Introduction

Environmental components

Geosphere / lithosphere

Solid earth, including soil, which supports most plant life

Biosphere

Living entities on Earth

Hydrosphere

Earth's water

ECOLOGY

The study of ecosystem

Atmosphere

Thin layer of gases that cover Earth's surface

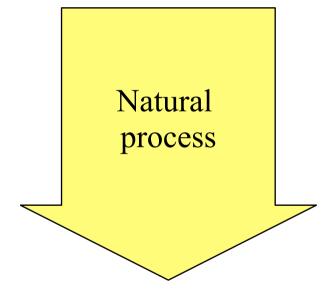
Ecosystem

- Encompasses both living (biotic) and non-living (abiotic) components of an area a combination of the community and physical and chemical components of the local environment.
- The major feature of this ecological level is the strong interaction between the biotic and abiotic components
- Major processes:
 - Nutrient recycling
 - Energy flow

Ecosystem processes

Energy flow:

- Energy sources
- Photosynthesis
- Primary production
- Secondary production

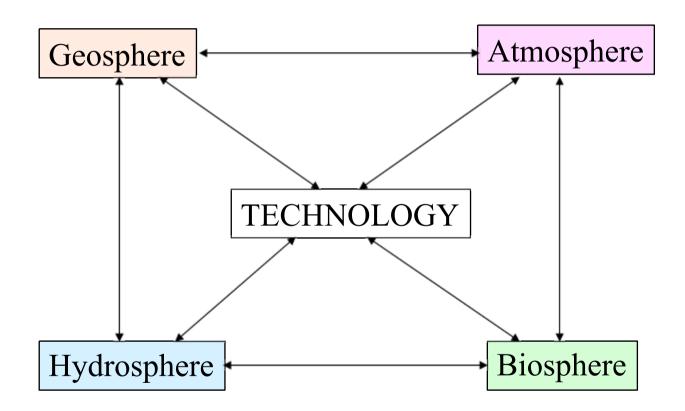


Major components & sub somponents

Environmental components

Abiotic	Hydrosphere (ocean, lake, river, groundwater)		
	Lithosphere (solid earth, soils)		
	Atmosphere		
Biotic	Living organisms (animal, plant, fungi, bacteria, virus)		
	Dead organic matters		

Environmental components

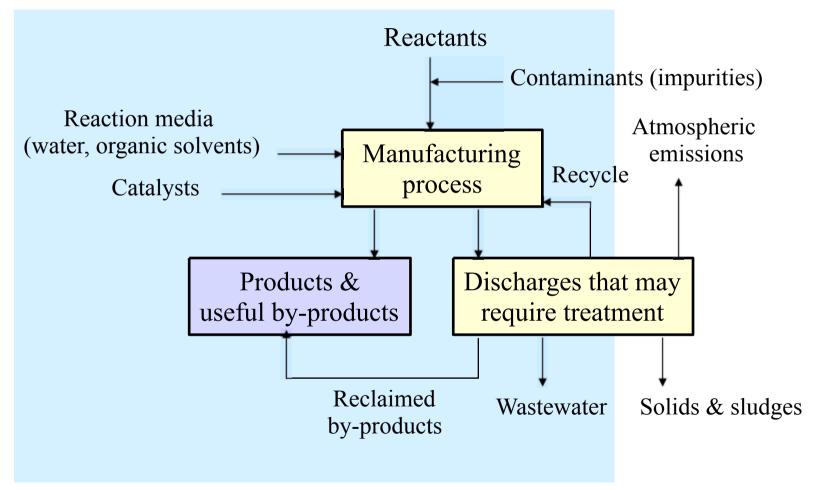


Multiple sources

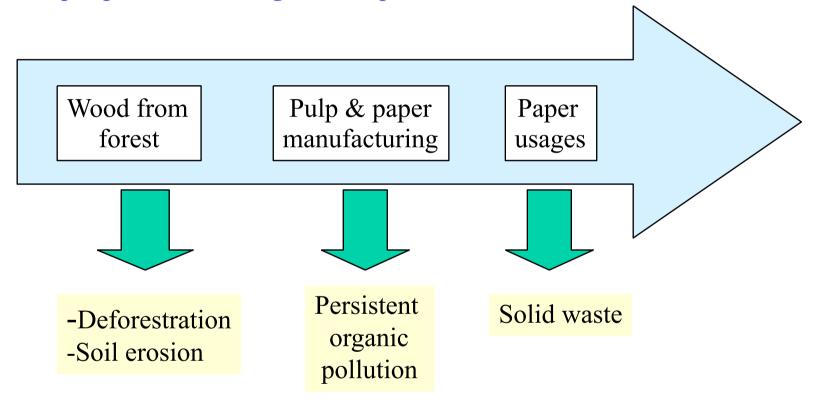
- Natural
 - -Volcano
 - -Dust
 - -CO2 from respiration and fire
- Man-made

Urbanisation and industrialisation enhanced pollution Degradation of self-purification processes

Manufacturing processes- from problems maker to problem solver

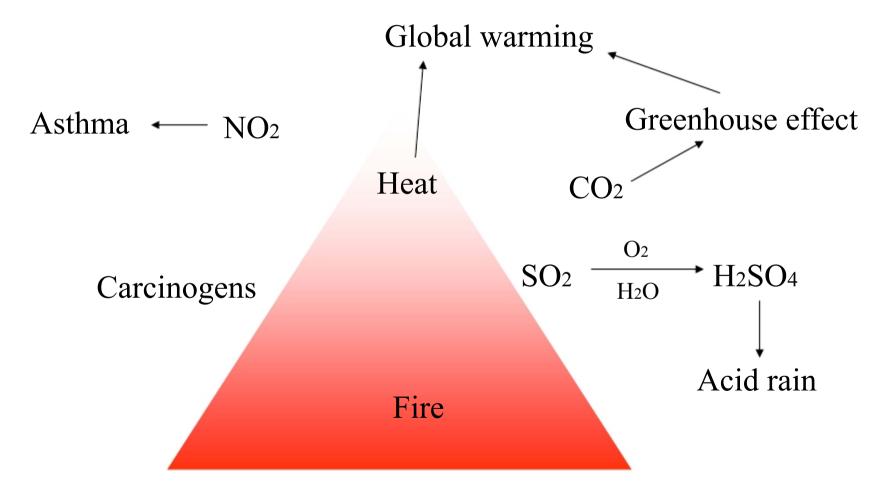


Paper production, usage and disposal

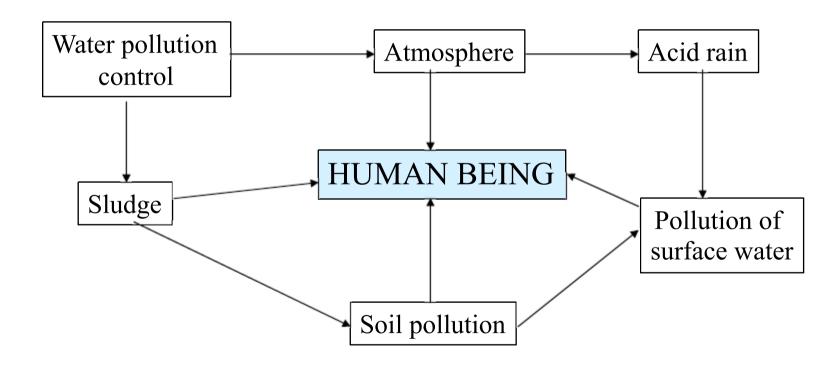


Environmental pollution generation

Burning of paper and other organic materials



Pollution from water to air to soil to water to air to soil to ...



Concept of tolerance

Each species tolerate to a range of optimum for physicochemical factors:

temperature

pН

light

nutrients

biological factors (food, competitors and predators)

Each species is most successful in that area where ranges of optima for different factors overlap to the greatest degree

Each species within a community and habitat has <u>different</u> and <u>unique niche</u>

Concept of disturbance

Discrete, punctuated killing, displacement or damaging of one or more individuals or colonies that directly or indirectly creates an opportunity for new individuals to be become established (Sousa, 1984)

Causes a temporary or permanent shift in the community

Risk assessment can be used to identify the risk after disturbances

Concept of pollution

Any change in the natural quality of the environment brought about by the following factors:

chemical physical biological

Normally, pollution causes by activities of man

Physical factors

change naturally in short term (flood, fire, storms, etc.) longer term change (e.g. climate change) man's activities (building, drainage, forest clearence)

Chemical factors

Changes through elevation of concentration of substances, e.g.

```
nutrients eutrophication
toxic substances health risk
organics reduce quality of raw water supply
```

Biological factors

Biological processes like predation or grazing, non-predatory effects like digging and man-induced events like tree felling, hunting etc. reduce species niche ecological imbalance cutting trees reduce oxygen generation capability

Note:

Eutrophication: Enrichment of nutrients in water bodies

Most pollution, disturbances can be recovered

The ability to recover and rate of recovery are dependent on the regime of disturbance:

Nature of disturbances

Size of the disturbed area

Magnitude and duration of the event (intensity of disturbing force)

Timing and frequency of the disturbance

Predictability of the disturbance

Turnover rate (average time required to disturb the entire area)

Major pollutants

Water, Atmosphere, Soil

Sulphur dioxide

Nitrogen oxides and nitrate

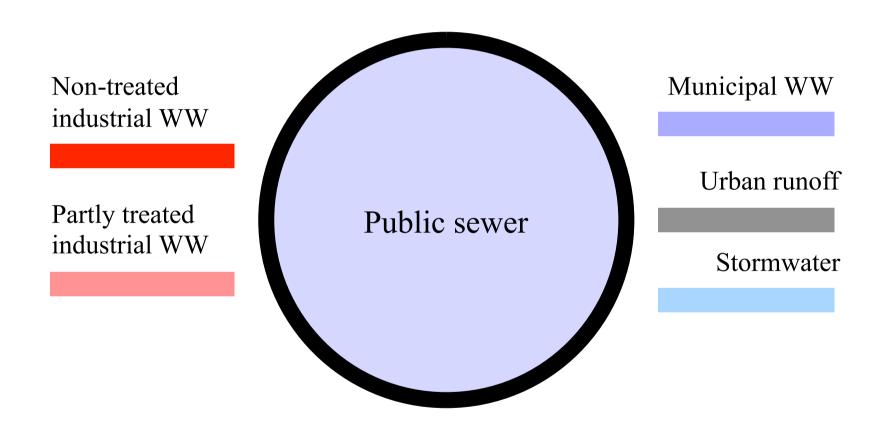
Sewage

Agricultural waste

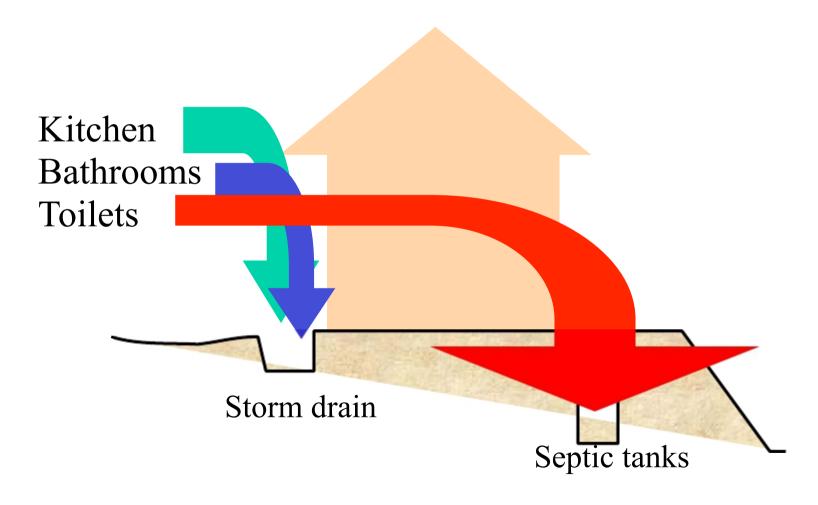
Warfare

Pesticides

Centralized (and combined) WWTP



Sewage in rural and remote areas

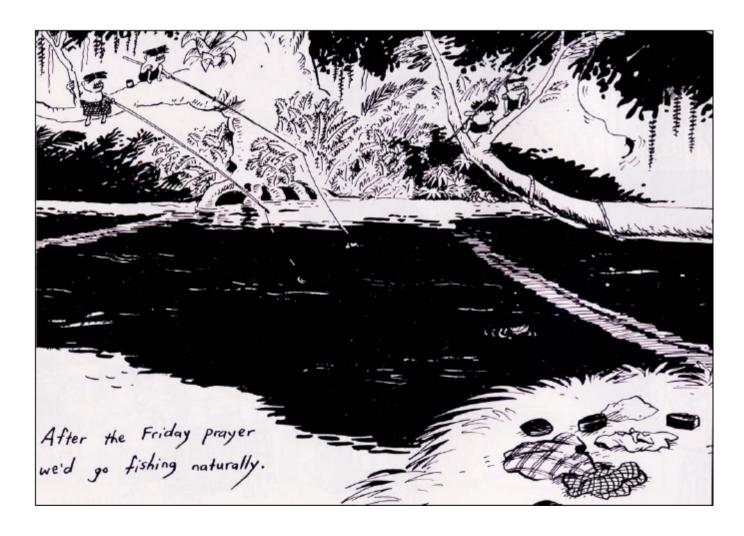


Major pollutants Water, Atmosphere, Soil

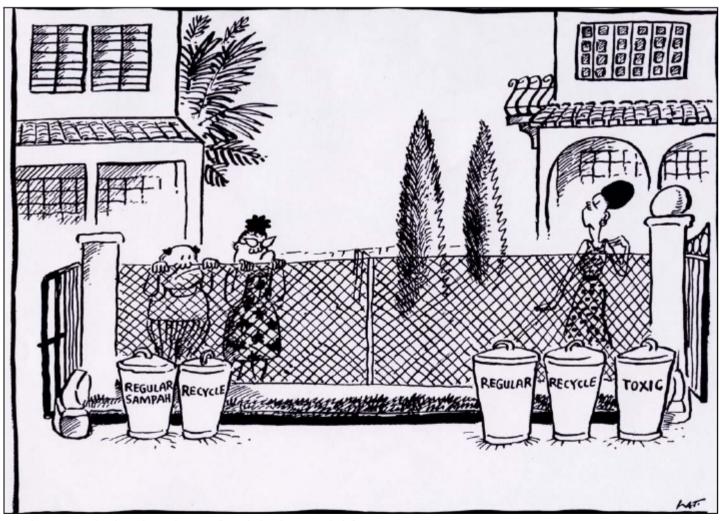


21

Major pollutants Water, Atmosphere, Soil



Major pollutants Water, Atmosphere, Soil



Lecture notes on Environmental Chemistry by Prof. Zaini Ujang. zaini@utm.my

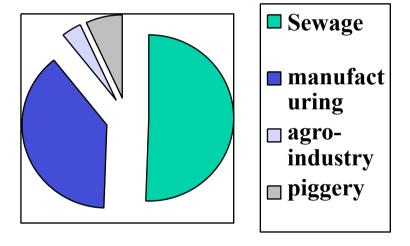
Number of water pollution sources by sector to Malaysian rivers (DOE, 2001)

Sewage plants (6,693)

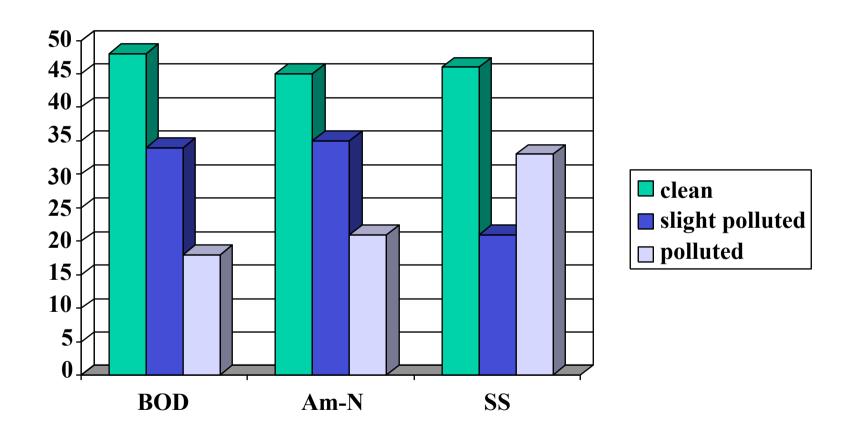
Manufacturing industries (5,086)

Pig farming (909)

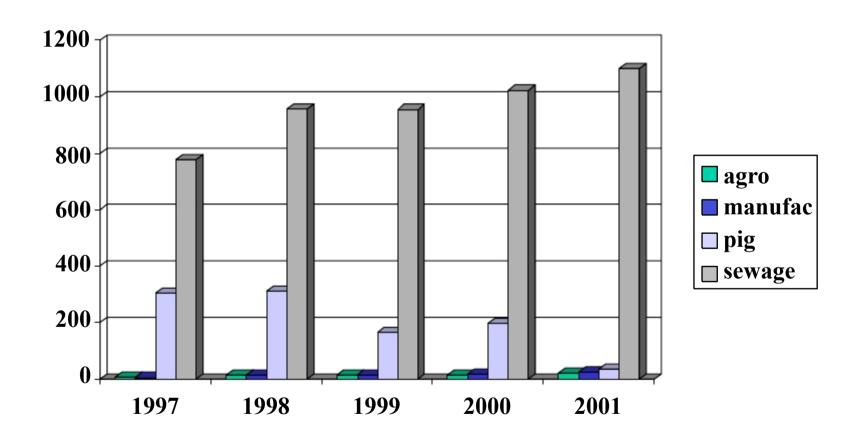
Agro-based industry (472)



Status of river basin water quality (DOE, 2002)



BOD loading by major sources, 1997-2001



Water pollution in perspective ...



Water pollution in perspective ...



Interim River Water Classes

	Classes				
Parameters	I	II	III	IV	V
Am-Nitrogen	0.1	0.3	0.9	2.7	>2.7
BOD	1	3	6	12	>12
COD	10	25	50	100	>100
DO	7	5-7	3-5	<3	<1
рН	6.5 - 8.5	6-9	5–9	5–9	-
Color (TCU)	15	150	-	-	-
TDS	500	1000	-	4000	-
TSS	25	50	150	300	>300
FC(per 100ml)	10	100	5000	5000	-
TC(per 100ml)	100	5000	20,000	50,000	>50,000

Pollution Distribution in Segget Catchment, JB

Locations	Contributors	Loading (Ton/d)	Percentage (%)
Before Rubbish	Industries	0	0
Trap	Sewage	0.9	26
	Non Pollution	0.4	11
	Sources		
Estuary	Industries	0	0
	Sewage	1.5	43
	Non Pollution	0.7	20
	Sources		
Total		3.5	100

Introduction to food webs

Some specific examples

Risk versus benefits

Development of safer chemicals

Introduction to food webs

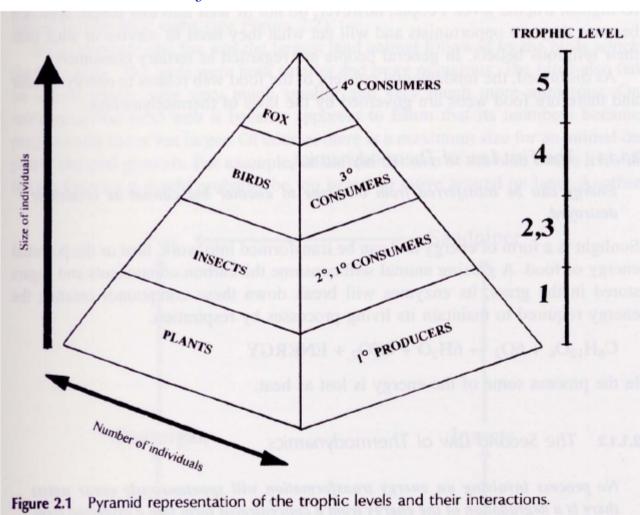
Food chains are dependent upon primary producers which input energy

Primary energy input is derived from photosynthesis (CO₂ converted to complex carbohydrates utilising sun's energy:

$$CO_2 + H_2O$$
 sunlight $CH_2O + O_2$

The problem: energy transfer is not efficient (energy loss as heat)

Introduction to food webs



Lecture notes on Environmental Chemistry by Prof. Zaini Ujang. zaini@utm.my

First law of thermodynamics

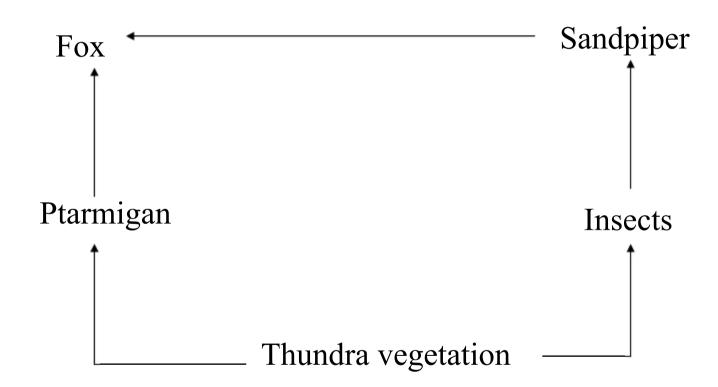
Energy can be transferred from one type to another but cannot be created or destroyed

$$C_6H_{12}O_6 + 6O_2 \rightarrow 6H_2O + 6CO_2 + Energy$$

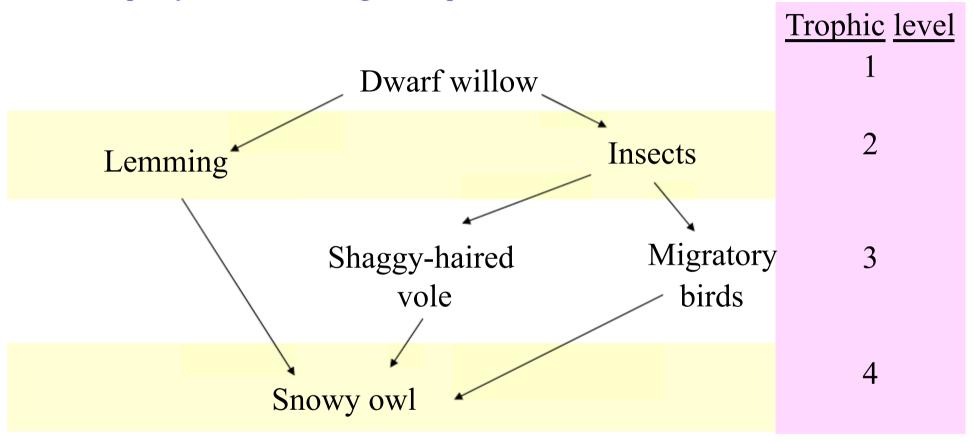
Second law of thermodynamics

No process involving an energy transformation will spontaneously occur unless there is a degradation of the energy from a concentrated form into a dispersed form

Schematic representation of interrelationship between 2 food chains



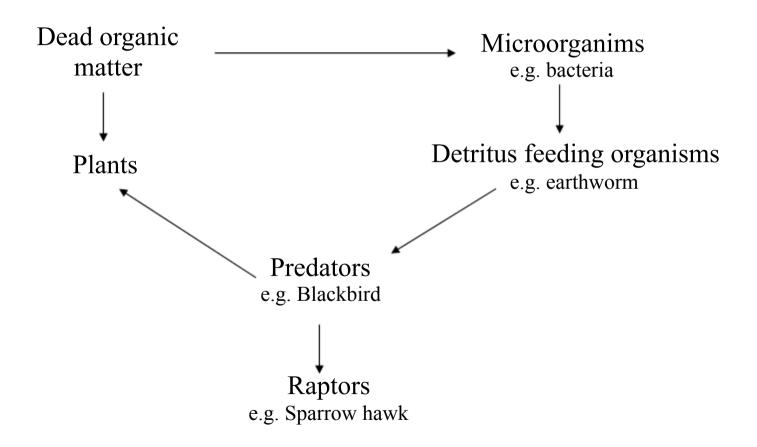
Complex food web showing the trophic levels



7 trophic levels showing examples of organisms

Trophic level	Description	Example
1 (P)	Primary producer	Spyrogyra, oak tree
2 (C1)	Primary consumer (herbivores)	Daphnia, elephant
3 (C2)	Secondary consumer (carnivores)	Water spider
4 (C3)	Tertiary consumer	Trout, wolf
5 (C4)	Quarternary consumer	Birds of prey
S	Saprophytes	Bacteria, fungi
D	Decomposers	Bacteria, earthworm

The use of dead organic matter



Fate & behaviour of chemicals in environment

E.g. Dichlorodiphenyltrichloroethane (DDT)

Very hydrophobic molecule which acts by interfering with ion transport systems in neuronal cell membrane

Inhibits neurotransmission → kills animals at certain dose

DDT (introduced in 1950s) is not species specific in its effects

DDT and related insecticides, endrin, dieldrin and aldrin are called ORGANOCHLORINE PESTICIDES

DDT revolutionised farming practices

DDT was developed by Swiss entomologist – Paul MÜller

Nobel Prize in 1948!

Banned in the mid-1960s in most developed countries

Fate & behavior of chemicals in environment

DDT in food chain in the USA

	Diet	DDT residues (ppm)		
Water	NA	0.0005		
Plankton	NA	0.04		
Sheepshead minnow	Plankton	0.94		
Pickeral	Predatory fish	1.33		
Heron	Small fish	3.57		
Herring gull	Scavenger	6.00		
Osprey (eggs)	Larger fish	13.8		
Merganser	Fish	22.8		
Cormorant	Larger fish	26.4		
Woodwelll et al. (1967) Science, 156, 821				

Risk versus benefits ...

Life is a risky business!

Concept of risk is complex

RISK = HAZARD x CHANCE (OF EXPOSURE)

Hazard = intrinsic property of a substance or an activity Occupational Safety and Health Act Risk associated with exposure to chemicals is recent Risk and benefit is not similar to all

Risk versus benefits ...



Is recycle is the solution to sustainable waste disposal and management?

Development of safer chemicals

Organochlorines (OCs) are dangerous to environment because of its indiscriminate toxicity

Environmental friendly chemicals?

E.g. pyrethroid insecticides

Pyrethrum is a mixture of several pyrethroids present in powdered *Chrysanthemum cinerariaefolium*, including pyrethrin, pyretol, pyrethrotoxic acid, pyrethrosin and chrysanthemine

Widely planted in Kenya

Pyrethroids act by modulating the gating characteristics of the sodium channel on neuronal membrane although the exact mechanism of the interaction between pyrethorid molecule and membrane sodium channel is not fully understood

Development of safer chemicals



Chemicals and products are to be disposed at high cost, or discharge to environment without proper treatment at all

Environmental toxicity testing

Toxicity testing in perspective

Extrapolating the results of toxicity test in rats (& other animals to humans)

The best is to use human body

Ethical problems, especially in pharmaceutical industry

Scope and limitations

Oestogenicity assay

Toxicity tests on animals and plants

Environmental impacts assessment

Environmental monitoring

Why monitor environmental contaminants

Methods

Meaning of analytical results

Analytical techniques

Identification of environmental contaminants

Inorganic contaminants

Immunoassays